### Markers of isoelectric point an essential tool in capillary isoelectric focusing

Filip Duša

Institute of Analytical Chemistry of the Czech Academy of Sciences





### Capillary isoelectric focusing a powerful separation method

- + Resolution as low as 0.01 pH unit
- Repeatability issues
  - pH gradient formation mixtures of hundreds of compounds (carrier ampholytes)
  - Surface quality/coating stability
  - Localized Joule heating → careful control of the total applied electric power
  - Two phases of the separation focusing and mobilization
  - Influence of sample matrix (e.g., salts, buffers)



# Internal standards of isoelectric focusing – isoelectric point (p/) markers



- Critical qualities of pl markers
  - Well-focusing ampholyte
    - ΔpK<sub>A</sub> < 4
  - Solubility at p/
  - Detectability at 280 nm
    - Absorption of carrier ampholytes below 280 nm
  - Observable at visual spectrum
  - Stability in solution



### A brief note on development of pl markers



#### Proteins

- Naturally occurring
- Have to be extracted from matrices
- High heterogeneity → isoforms, posttranslational modifications
- Low stability against degradation



#### Peptides

- Based on tryptophan (absorption at 280 nm)
- Designed and synthesized on demand (Shimura et al., 2000 →)
- High purity
- Low heterogeneity
- Low stability at ambient temperature



### Low-molecular-mass (LMM) p/ markers



- Simple well-absorbing core, that can be easily modified by attaching functional groups
- Preferably one-step synthesis
- Fulfilling pl marker requirements
- First development and synthesis from 1994\* and continues till present
- Universality, selectivity, sensitivity



#### *Universality* Nitrophenol based LMM pl markers (NPIM)



- Yellow color of the pl markers\*
- $\lambda_{MAX} = 400 500 \text{ nm} \rightarrow \text{visual detection}$
- Low molecular mass  $\rightarrow$  fast focusing
- High stability in powder and in solution
- 11 NPIMs calibrated using commercially available peptide pI markers
- MS detectable\*\*



\*Duša, F., Moravcová, D., Šlais, K. Anal. Chim. Acta 2019, 1076, 144–153. \*\*Mazanec, K., Bobalová, J., Šlais, K. Anal. Bioanal. Chem. 2009, 393, 1769–78.

# 41 NPIMs developed and characterized

- 3.16 10.36 pH range covered
- pH coverage gaps identified
  - 9.77 8.43
  - 5.95 5.29
  - 4.55 3.93
- Hard to distinguish individual markers



#### *Selectivity* Colored p*I* markers (CPIM)

- Further UV-Vis absorbing cores added
  - Azo dyes based PIMs 41 compounds
  - Sulfonephthalein/Phthalein based PIMs
    - 17 compounds
  - Phenol based PIMs (UV absorption only)
    - 14 compounds





**iac** 

## Differently absorbing p/ markers enable direct identification





Duša, F., Moravcová, D., Šlais, K. Anal. Chim. Acta 2022, 1221, 340035.

#### Coelution of PIM and analyte



- Example of coeluting compounds
  - peptide p/ 5.50 ( $\lambda_{MAX}$  = 280 nm)
  - CPIM 5.52 ( $\lambda_{MAX}$  = 378 nm)
- Both compounds can be easily distinguished using 2 detection wavelengths
- fluctuations of pH gradient are diminished



### pH gradient densely covered by the sum of PIMs



- Gaps present in the NPIM range were filled by CPIM
- Appropriate combination enabled to selectively determine each marker in the mixture
- 113 compounds available for tracing pH gradient



#### *Sensitivity* Fluorescent p/ markers\*





\*Dadajová, P., Čmelík, R., Šlais, K., Duša, F., soon to be published data

### Purity of the FPIMs

- Problematic synthesis
  - Contamination by byproducts
  - Low fluorescence of products
- cIEF-UV and cIEF-LIF analysis
  - 65 initial candidate structures
  - 19 FPIMs passed
    - purity (at least 50 % intensity of the main peak)
    - fluorescence intensity (0.5 peak area RFU)
- MS analysis was used to confirm the expected masses of the FPIMs passing the criteria







#### FPIMs Coverage of pH gradient

- Most of the compounds had pls at neutral and acidic region
- Only a few FPIMs available for marking the basic pH
- None of the FPIM above the p/ 8.73 passed the 50 % purity criterium
  - The criterium lowered to 40 % to add two more basic FPIMs









 The final product of the FPIM synthesis contained impurities at pH range 9.4 – 10.2

Purification:

- 75  $\mu$ m i.d. capillary
- Lysine (p/ 9.7) spacer to separate the impurities in cIEF
- 9 fractions collected to chemical mobilizer solution (HAc) and vacuum-evaporated



Method:

Sciex PACE MDQ Plus; LPA coated 75  $\mu$ m i.d. capillary,  $I_{tot}$  = 30 cm,  $I_{eff}$  = 20 cm; focusing 17 min at 20 kV; mobilization 11 min at 20 kV; fractionation 9 times 2 min at 20 kV; detection: argon-ion laser 488 nm, notch filter 488 nm / bandpass filter 520/20 nm. Sample:

1.92 M Urea, 64 % V/V cIEF gel, 0,64 % w/V Pharmalyte 3-10, 120 mM L-Arginine, 1.6 mM Iminodiacetic acid, 15 ng mL<sup>-1</sup> FPIM 10.21

# Universality, Selectivity, and Sensitivity LMM p/ markers



#### Total of 134 compounds for tracing pH gradient



# Institute of Analytical Chemistry **Isoelectric focusing markers group** • Pavlína Dadajová Dana Moravcová $\bullet$ Richard Čmelík • Karel Šlais

#### **NPIM 7.54** NPIM 6.737 + NPIM 6.735 **NPIM 7.24 NPIM 7.56** NPIM D NPIM m 7.37 **NPIM 6 95** 6.83 **NPIM 7.84** NPIM NPIM 6 3 6.39 **NPIM 6.60** NPIM 8 mo NPIM **NPIM 6.21 NPIM 8.43** 7.23+7. NPIM 6.083 + NPIM 6.077 **NPIM 5.95** ma NPIM 8.83 NPIM 5.299 **NPIM 5 295** ų NPIM 5. NPIM 5.0 **NPIM 9.77** NPIM 4 **NPIM 10.01** NPIM 10.19 NPIM 10.36 NPIM 4.55 + NPIM 4.56 + NPIM 10.21 xcluded NPIM 2.78 **NPIM 4.97** NPIM 3.93 **NPIM 3.56** $\Omega \Omega$ **NPIM 4 97** \*\* NPIM 4.94 absorbing NPIM peak at p/ 3.88

#### Thank you for attention!